Acta Crystallographica Section E

## Structure Reports

Online
ISSN 1600-5368

## Holger Gritzo, Frank Schaper and Hans-Herbert Brintzinger*

Fachbereich Chemie, Universität Konstanz, Postfach 5560, 78434 Konstanz, Germany

Correspondence e-mail:
hans.brintzinger@uni-konstanz.de

## Key indicators

Single-crystal X-ray study
$T=188 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.004 \AA$
$R$ factor $=0.070$
$w R$ factor $=0.190$
Data-to-parameter ratio $=18.2$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
(C) 2004 International Union of Crystallography Printed in Great Britain - all rights reserved

## $\operatorname{Bis}\left(\boldsymbol{\eta}^{3}\right.$-indenyl)bis(tetrahydrofuran)magnesium(II)

In the crystal structure of the title compound, $\left[\mathrm{Mg}\left(\mathrm{C}_{4} \mathrm{H}_{8} \mathrm{O}\right)_{2}\left(\mathrm{C}_{9} \mathrm{H}_{7}\right)_{2}\right]$, the two indenyl ligands are bound to the Mg atom in a manner intermediate between $\eta^{1}$ - and $\eta^{3}$ coordination; the latter is untypical, as it includes one of the ring-sharing C atoms. The $\mathrm{O}-\mathrm{Mg}-\mathrm{O}$ angle $\left[101.01^{\circ}\right.$ ] is larger than for any other magnesocene-(THF) $2_{2}$ adduct (THF is tetrahydrofuran). A crystallographic $C_{2}$ axis bisects the $\mathrm{O}-$ $\mathrm{Mg}-\mathrm{O}$ angle.

## Comment

The title compound, (I), was prepared as a precursor for syntheses of other metallocenes. (Ind) $)_{2} \mathrm{Mg}$ is known to be polymeric in the solid state, with indenyl (Ind) groups coordinated in monohapto, dihapto or pentahapto arrangements to one or two Mg centres (Atwood \& Smith, 1974). By addition of tetrahydrofuran (THF) to this complex, which was prepared via a ligand-exchange reaction (Eisch \& Sanchez, 1985), we obtained its THF adduct, (I), which is monomeric in the solid state.

(I)

A crystallographic $C_{2}$ axis bisects the $\mathrm{O} 1-\mathrm{Mg}-\mathrm{O} 1^{\mathrm{i}}$ angle (symmetry code as in Table 1) and the two indenyl ligands are thus equivalent by symmetry. Unusual features of the coordination geometry of (I) are revealed by comparison with other magnesocene structures. The shortest $\mathrm{Mg}-\mathrm{C}(\mathrm{Cp})$ distance ( Cp is cyclopentadienyl) of 2.256 (3) $\AA$ in (I) is similar to the bona fide $\mathrm{Mg}-\mathrm{C}\left(\eta^{1}\right)$ distance of 2.282 (2) $\AA$ for the related compound $\left(\eta^{1}-\mathrm{Cp}\right)\left(\eta^{5}-\mathrm{Cp}\right) \mathrm{Mg}(\mathrm{THF})_{2}$, (II) (Jaenschke et al., 2003). The next two $\mathrm{Mg}-\mathrm{C}$ distances in increasing length are 2.723 (3) and 2.738 (3) $\AA$ for (I), compared with 2.736 (2) and 2.880 (2) $\AA$ for (II). The shortening of one of these $\mathrm{Mg}-\mathrm{C}$ bond distances indicates that the hapticity of the indenyl ligands in (I) is slightly higher than the $\eta^{1}$-coordination assigned to (II). In (I), one of the three

Received 19 May 2004
Accepted 1 July 2004
Online 17 July 2004
$\qquad$


Figure 1
The molecular structure of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the $50 \%$ probability level and H atoms have been omitted for clarity. Suffix $A$ corresponds to symmetry code (i) in Table 1.
closest-bound C atoms is in a ring-sharing position of the indenyl ligand.

In another comparable structure, viz. $\mathrm{C}_{2} \mathrm{H}_{4}\left(\eta^{3} \text {-Ind }\right)_{2}-$ $\operatorname{Mg}(\mathrm{THF})_{2}$, (III) (Damrau et al., 1998), one of the $\eta^{3}$-coordinated indenyl ligands has one of its Mg -bound C atoms in a ring-sharing position, as in (I). The three shortest $\mathrm{Mg}-\mathrm{C}$ distances in (III) $[2.328$ (3), 2.451 (3) and 2.682 (3) $\AA$ ] are closer to each other than those in (I). This and their smaller average indicate that the hapticity of the indenyl ligands in (I) is lower than their $\eta^{3}$-coordination in (III). Corresponding C C distances of the indenyl fragments are quite similar for (I) and (III); however, the maximum difference is $0.02 \AA$.

Normally, an allylic $\eta^{3}$-coordination of an indenyl ligand involves those three C atoms which are not part of the $\mathrm{C}_{6}$ ring (O’Connor \& Casey, 1987). Coordination of an indenyl fragment to an Mg atom involving a ring-sharing C atom, as in (I), has so far been seen only in two bridged magnesocenes [Cambridge Structural Database (CSD, Version 5.25 of November 2003) refcodes PUBREY and PUBROT; Allen, 2002].

The $\mathrm{Mg}-\mathrm{O}$ distance of the magnesium-THF fragments $[\mathrm{Mg}-\mathrm{O}=1.993(2) \AA$ ] is comparable with those found for other magnesocene-THF adducts $(\mathrm{Mg}-\mathrm{O}=1.989-2.098 \AA$ (CSD refcodes HUXSAU, NUZXIP, NUZXOF, PUBREJ, PUBRIN, PUBROT, PUBRUZ, ROBCUG, TACXIF, VITZUT, WIDYAJ, WIDYEN, XILZIB and ZEHYUG). The $\mathrm{O} 1-\mathrm{Mg}-\mathrm{O} 1^{\mathrm{i}}$ angle in (I) $\left[101.01(12)^{\circ}\right]$ is the largest observed so far for any magnesocene with two THF ligands $\left(\mathrm{O}-\mathrm{Mg}-\mathrm{O}=88.20-94.40^{\circ} ; ~ \mathrm{CSD}\right.$ refcodes HUXSAU,

NUZXIP, PUBREJ, PUBROT, PUBRUZ and XILZIB). This opening of the $\mathrm{O}-\mathrm{Mg}-\mathrm{O}$ angle is probably connected with the unusually low hapticity of both indenyl ligands in (I).

## Experimental

All manipulations were performed using conventional Schlenk techniques under an atmosphere of argon or in a glove-box under an atmosphere of nitrogen. (Ind) ${ }_{2} \mathrm{Mg}$ (Eisch \& Sanchez, 1985) (5.0 g, 20 mmol ) was stirred in a mixture of 50 ml pentane and 30 ml tetrahydrofuran for 3 d at room temperature. The yellow suspension was decanted and the precipitate washed twice with 20 ml pentane. Drying in vacuo yielded 6.4 g ( $16 \mathrm{mmol}, 82 \%$ ) of a pale-yellow powder. For crystallization, a small amount was dissolved in tetrahydrofuran. While the solvent was slowly evaporated in a glove-box at room temperature colourless crystals were formed. ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CD}_{2} \mathrm{Cl}_{2}, 298 \mathrm{~K}$ ): $\delta 7.54(4 \mathrm{H}, m$, Ind-H6), $6.89(4 \mathrm{H}, m$, IndH7), 6.79 ( $2 \mathrm{H}, s$, Ind-H3), $5.85(4 \mathrm{H}, s$, Ind-H2), 3.18 ( $8 \mathrm{H}, b s$, THF), 1.71 ( $8 \mathrm{H}, b s$, THF). ${ }^{13} \mathrm{C}$ NMR ( $600 \mathrm{MHz}, \mathrm{CD}_{2} \mathrm{Cl}_{2}, 298 \mathrm{~K}$ ): $\delta 133.9$ (Ind-C1), 120.6 (Ind-C3), 120.2 (Ind-C6), 117.9 (Ind-C7), 89.5 (IndC2), 70.4 (THF), 25.3 (THF). Analysis calculated for $\mathrm{C}_{26} \mathrm{H}_{30} \mathrm{MgO}_{2}$ : C 78.30, H 7.58\%; found: C 76.68, H 7.65\%.

## Crystal data

$\left[\mathrm{Mg}\left(\mathrm{C}_{4} \mathrm{H}_{8} \mathrm{O}\right)_{2}\left(\mathrm{C}_{9} \mathrm{H}_{7}\right)_{2}\right]$
$M_{r}=398.81$
Monoclinic, $C 2 / c$
$a=10.958(4) \AA$
$b=9.683$ (6) $\AA$
$c=20.861$ (9) $\AA$
$\beta=96.56(5)^{\circ}$
$V=2199.0(18) \AA^{3}$
$Z=4$
$D_{x}=1.205 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 25
reflections
$\theta=5.1-12.5^{\circ}$
$\mu=0.10 \mathrm{~mm}^{-1}$
$T=188 \mathrm{~K}$
Plate, colourless

$$
0.4 \times 0.2 \times 0.2 \mathrm{~mm}
$$

## Data collection

Bruker $P 4$ diffractometer $\omega$ scans
5253 measured reflections
2399 independent reflections
1658 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.077$
$\theta_{\text {max }}=27.0^{\circ}$

$$
\begin{aligned}
& h=-13 \rightarrow 12 \\
& k=-12 \rightarrow 12 \\
& l=-26 \rightarrow 26
\end{aligned}
$$

3 standard reflections every 97 reflections intensity decay: $0.2 \%$

## Refinement

Refinement on $F^{2}$

$$
\begin{aligned}
& \begin{array}{c}
w=1 /\left[\sigma^{2}\left(F_{o}{ }^{2}\right)+(0.0986 P)^{2}\right. \\
\quad+1.492 P] \\
\text { where } P=\left(F_{o}{ }^{2}+2 F_{c}^{2}\right) / 3 \\
(\Delta / \sigma)_{\max }<0.001 \\
\Delta \rho_{\max }=0.30 \mathrm{e}^{2} \AA^{-3} \\
\Delta \rho_{\min }=
\end{array}-0.73 \mathrm{e}^{-3}
\end{aligned}
$$

$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.070$
$w R\left(F^{2}\right)=0.190$
$S=1.04$
2399 reflections
132 parameters
H -atom parameters constrained

## Table 1

Selected geometric parameters ( $\AA{ }^{\circ}{ }^{\circ}$ ).

| $\mathrm{Mg} 1-\mathrm{O} 1$ | $1.993(2)$ | $\mathrm{C} 2-\mathrm{C} 3$ | $1.429(4)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{Mg} 1-\mathrm{C} 2$ | $2.256(3)$ | $\mathrm{C} 3-\mathrm{C} 4$ | $1.391(4)$ |
| $\mathrm{Mg} 1-\mathrm{C} 3$ | $2.723(3)$ | $\mathrm{C} 4-\mathrm{C} 5$ | $1.423(4)$ |
| $\mathrm{Mg} 1-\mathrm{C} 1$ | $2.738(3)$ | $\mathrm{C} 5-\mathrm{C} 6$ | $1.412(4)$ |
| $\mathrm{C} 1-\mathrm{C} 9$ | $1.408(4)$ | $\mathrm{C} 6-\mathrm{C} 7$ | $1.361(4)$ |
| $\mathrm{C} 1-\mathrm{C} 2$ | $1.441(4)$ | $\mathrm{C} 7-\mathrm{C} 8$ | $1.418(5)$ |
| $\mathrm{C} 1-\mathrm{C} 5$ | $1.447(4)$ | $\mathrm{C} 8-\mathrm{C} 9$ | $1.378(4)$ |

$\mathrm{O} 1-\mathrm{Mg} 1-\mathrm{Ol}^{\mathrm{i}} \quad 101.01$ (12)
Symmetry code: (i) $-x, y, \frac{3}{2}-z$.
H atoms were positioned geometrically and refined as riding, with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}$ (parent atom).

## metal-organic papers

Data collection: XSCANS (Siemens, 1992); cell refinement: $X S C A N S$; data reduction: $X S C A N S$; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Siemens, 1997); software used to prepare material for publication: SHELXTL.

## References

Allen, F. H. (2002). Acta Cryst. B58, 380-388.
Atwood, J. L. \& Smith, K. D. (1974). J. Am. Chem. Soc. 96, 994-998.

Damrau, H. R. H., Geyer, A., Prosenc, M. H., Weeber, A., Schaper, F. \& Brintzinger, H. H. (1998). J. Organomet. Chem. 553, 331-343.
Eisch, J. J. \& Sanchez, R. (1985). J. Organomet. Chem. 296, C27C31.
Jaenschke, A., Paap, J. \& Behrens, U. (2003). Organometallics, 22, 11671169.

O’Connor, J. M. \& Casey, C. P. (1987). Chem. Rev. 87, 307-318.
Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
Siemens (1992). XSCANS. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
Siemens (1997). SHELXTL. Version 5.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.

